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 $CvO_2 = (Hb \times 1.34 \times SvO_2) + (PvO_2 \times 0.003)$  Shunt Equation ( $Q_s/Q_t$ )  
 $Q_s/Q_t = (CcO_2 - CaO_2) / (CcO_2 - CvO_2)$  Modified Shunt Equation ( $Q_s/Q_t$ )  
 $Q_s/Q_t = ((PAO_2 - PaO_2) \times 0.003) / ((CaO_2 - CvO_2) + (PAO_2 - PaO_2) \times 0.003)$

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C THREE CALCULATIONS WITH CHEMICAL FORMULAS ...Part Two: Determining Chemical Formulas A. Percent

Composition From Compound Formulas. (Section 3.3)

1. If Formula Of Compound Is Known, Its Percent Composition By Mass Of Each Element Can Be Found, And Vice Versa.  $\div$  Atomic Masses (AM) Compound Percent Composition Formula In Terms Of Elements X Atomic Masses (AM) 2. Jun 6th, 2022

Chapter 9 Chemical Calculations And Chemical Formulas 118

Study Guide For An Introduction To Chemistry Section

9.2 Relating Mass To Number Of Particles Goals To

Show How To Do A Procedure Called Counting By Weighing. To Introduce Atomic Mass And Show How It Can Be Used To Convert Between The Mass Of A

Sample Of An Element And The Number Of Atoms That The Sampl Feb 18th, 2022

Calculations With Chemical Formulas And Equations With Chemical Formulas And Equations 2 | 1 . 2 ... Percentage Composition

Example: A Sample Of A Liquid With A Mass Of 8.657 G Was ...

•Smallest Is  $3.722 \times 10^{-3}$  Mole •C = •H =

Step 3: Divide All Number Of Moles By The Smallest Feb 3th, 2022.

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 $D_x e^x = e^x$  (14)  $D_x a^x = a^x \ln a$  (15)  $D_x \ln|x| = \frac{1}{x}$  (16)  $D_x \sin^{-1} x = \frac{1}{\sqrt{1-x^2}}$  (17)  $D_x \cos^{-1} x = -\frac{1}{\sqrt{1-x^2}}$  (18)  $D_x \tan^{-1} x = \frac{1}{x^2 + 1}$  (19)  $D_x \cot^{-1} x = -\frac{1}{x^2 + 1}$  (20) D May 16th, 2022.

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