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\* $\text{LiNO}_3 + \text{PbSO}_4 + \text{C}_3\text{H}_8 + \text{CO}_2$  1-120  $\text{Zn} + \text{HCl} + \text{H}_2\text{O}$  10. 11. 12. ...Types Of  
Reactions Worksheet Balance The Following Equations And Indicate The Type Of  
Reaction Taking Place:  $\text{H}_3\text{PO}_4 + \text{HBr} + \text{Na}_3\text{PO}_4 + \text{NaBr} + \text{Al}_2(\text{SO}_4)_3$  Type Of Reaction:  
 $\text{CaSO}_4 + \text{Mg} + \text{C}_2\text{H}_4 + \text{PbSO}_4 + \text{NH}_3 + \text{H}_2\text{O} + \text{O}_2 + \text{PbSO}_3 + \text{MgO} + \text{Fe}$   
+ Type Of Reaction:  $\text{H}_2\text{O} + \text{CO}_2$  + Type Of Reaction: T Feb 17th, 2022

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